CO10787-00053

Region 8 9042

70757 200053 - PERMIT, NON

PART I. AUTHORIZATION TO INJECT

Pursuant to the Underground Injection Control Regulations of the U.S. Environmental Protection Agency codified at Title 40 of the Code of Federal Regulations, Parts 124, 144, 146 and 147,

SHELL WESTERN E&P INC. Subsidiary of Shell Oil Company Post Office Box 576 Houston, Texas 77001

is hereby authorized to operate a Class I injection well, commonly known as Hovenweep (HWD-1) located at 300 feet from South line and 515 feet from East line of Section 9, Township 38N, Range 18W of Montezuma County, Colorado. Injection shall be for the purpose of industrial waste fluid disposal into the Leadville Limestone and Ouray Formation, in accordance with conditions set forth herein.

Permit authorization shall not commence until the operator has fulfilled and verified with the Director all applicable conditions of this permit. "Transition from Rule to Permit Authorization" requirements are set forth in Part II, Section C. 1. of this permit.

All conditions set forth herein refer to Title 40 Parts 144, 146, and 147 of the Code of Federal Regulations and are regulations that are in effect on the date that this permit is effective.

This permit consists of a total of 30 pages and includes all items listed in the Table of Contents. Further, it is based upon representations made by the permittee and on other information contained or referenced in the Administrative Record. It is the permittee's responsibility to read and understand all provisions of this permit.

This permit and the authorization to inject are issued for ten (10) years, unless terminated. The permit will be reviewed by EPA at least every five years to determine whether action under 40 CFR 144.36 (a) is warranted. The permit will expire at midnight February 6, 1997, or upon delegation of primary enforcement responsibility for the UIC-1422 Program to the State of Colorado, unless that State has adequate authority and chooses to adopt and enforce this permit as a State permit.

Issued this 6th day of February , 1987.

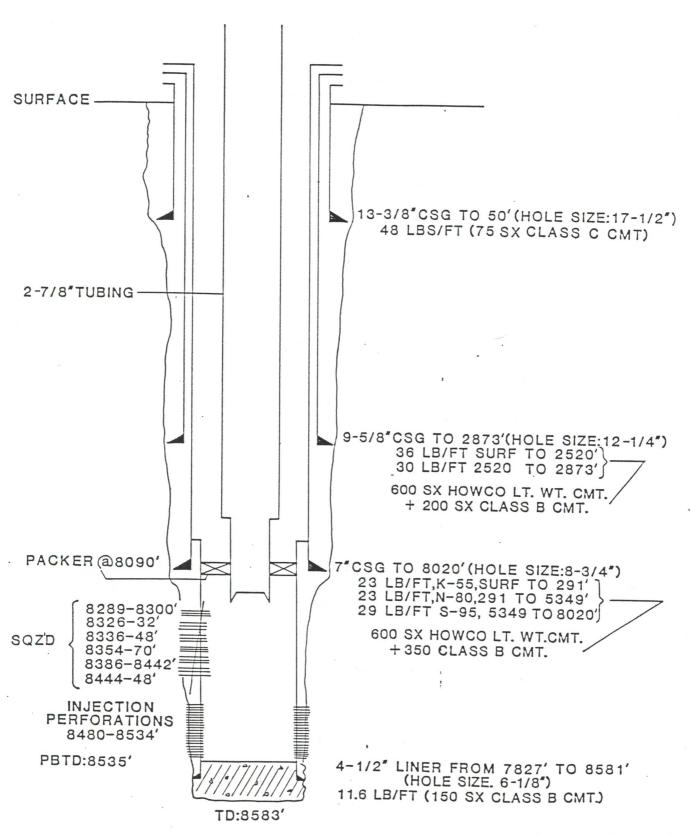
This permit becomes effective March 8, 1987

Max H. Dodson, Director Water Management Division *

* NOTE: The person holding this title is referred to as the "Director" throughout this permit.

Page 4 of 30 EPA Permit No. COS12ME-0042

CONSTRUCTION DETAILS OF HWD-1



Page 20 of 30 EPA Permit No. COS12ME-0042

Page 3 of 14 Hovenweep (HWD-1)

The draft permit contains a condition which states that EPA may with due cause, modify, revoke and reissue, or terminate the permit in accordance with Federal regulations, including if and when revisions or amendments to the Safe Drinking Water Act (SDWA) are made.

g. <u>Injection Zone Hydrogeology</u>

The injection zone is in the Leadville Limestone of Mississippian age and the Devonian age Ouray Formation The Leadville Limestone occurs at a depth of 8,230 feet from surface and is 284 feet thick. The Ouray Formation is at 8,514 feet from surface and is not penetrated through the entire formation (TD is 8,583 feet; PBTD is 8,535 feet). Lithology of the Leadville Limestone is limestone, often oolitic and fossiliferous, changing to dolomite in the lower half of the unit. The Ouray Formation is limestone and dolomite with occasional streaks of gray-green waxy shale.

Both formations are perforated in the interval of 8,480-8,534 feet from surface. Hydrogeologic parameters for this interval are:

Porosity = 8%

Permeability = 2.7 md

Fluid pressure = 2,500 psi

Fracture pressure = 4,330 psi

Bottom hole pressure = 3,220 psi at 3 BPM

Bottom Hole temperature = 180° F

Total Dissolved Solids (TDS) ranges from 3,000 - 100,000 ppm,

depending on the location of the gas water interface ("Free water" is the higher TDS value).

h. Confining Zone Hydrogeology

The overlying Molas Formation has been described as the confining zone for the Woods #3 (MWD-1) well which is also in the McElmo Dome Field. Because of the consistancy in geology accross the McElmo Dome field, the Molas Formation shall be considered the confining zone for the entire Hovenweep well as well. The Molas Formation is a Pennsylvanian age redbed, consisting of interbedded red siltstone and sandstone, light colored limestone and varicolored shales. This formation was evaluated by electric logs to estimate its petrophysical properties. These logs included a Borehole Compensated Sonic Log, a Compensated Neutron Log, a Formation Density Log, a Duel Laterolog and microlaterolog-microlog. The interpretation of these logs showed the formation to be approximately 114 feet thick. The shales and siltstones of the Molas Formation have apparent log porosity but are impermeable. The limestone intervals are tight, with porosity less than two percent.

Other shale formations between the injection zone and the lowermost possible USDW include the Cutler Formation, the Moenkopi Formation and the Chinle Formation. These three units are considered redbeds and have low permeabilities. The Chinle Formation is 706' thick, the Moenkopi Formation is 49' thick and the Cutler Formation is 1,777' thick. Calcualtions of TDS concentrations using resistivity logs have shown the Chinle and Cutler Formations not to be USDW's, although they are water bearing and produce water elsewhere in the region.

i. Local Groundwater Utilization

Except for this facility, there are no known wells penetrating the injection zone within the area of review. The deepest water well (680 feet) is located in Township 38N, Range 18W. Water production is from the Dakota Sandstone and possibly the Salt Wash Member, of the Morrison Formation. Other potential fresh water bearing units were identified, in the report titled Mineral and Water Resources of Colorado in Water Resources, by J.W. Odell, D.L. Coffin, and R.H. Langford (1964), as "... in order of importance are, the Dakota Sandstone, of Cretaceous age. The Entrada and Junction Creek Sandstone and the Salt Wash Member of the Morrison Formation, all of Jurassic age; and the Wingate Sandstone of Triassic age. Older rocks generally yield no water or water that is too highly mineralized for ordinary uses."

The report titled Atlas of Ground Water Quality in Colorado by F.N. Repplier, E.C. Healy, D.B. Collins and P.A. Longmire (1981), shows ground water production from the Cretaceous age Dakota Sandstone with water quality of 1,410 ppm TDS. This water well is over 10 miles to the east of the disposal well. There is also water production from the Jurassic age Morrison Formation, of 1,450 ppm TDS water, 15 miles to the south. All of the known and potential water bearing formations are hydrogeologically isolated from the injection zone by 1) the identified confining layer, 2) cement in the wellbore, and 3) approximately 2,000 feet of other redbeds and shale.

John Romero, Supervisory Water Resource Engineer, for the Colorado Department of Natural Resources, researched the water resources of the subsurface around the McElmo Dome Unit. Known or suspected aquifers in the area are surfical deposits, the Dakota and Burro Canyon Formations, Salt Wash member of the Morrison Formation, the Junction Creek and Entrada Sandstone and (possibly) the Chinle Formation. The underlying Cutler and Rico Formations have aquifer potential but definitive data are lacking, (Romero, 1985). However, as discussed in Section 10. a. 1. of this Fact Sheet, estimates of TDS in the Cutler indicates that the Cutler is not a USDW.

PHYSICAL SETTING

a. Structural Geology

The Hovenweep (HWD-1) well is located on the eastern flank of the Colorado Plateau physiographic province. This province is characterized by thick sequences of sediments which have been structurally stable since Precambrian time. Major geologic events, in more recent time (Tertiary), include the intrusion of laccolithic stocks such as the Sleeping Ute Mountain. This volcanic/plutonic rock body provided the heat source that caused the Leadville Limestone to alter and produce $\rm CO_2$ gas, which accumulated in economic quantities in the McElmo Dome area.

Page 5 of 14 Hovenweep (HWD-1)

b. Stratigraphy

The Colorado Plateau is characterized by thick sedimentary sequences and a tectonically stable environment since Precambrian time. The stratigraphic units, penetrated by the Hovenweep (HWD-1) well, are shown in the appendix to this Fact Sheet.

c. Hydrogeology

Hydrogeologic horizons have been identified (in the reports referenced or included in the Administrative Record) to be surficial deposits which are normally less than 20 feet thick except along valley bottoms. No water wells utilize these types of deposits in the area, (Romero, 1985). "Dakota Sandstone, Burro Canyon, and Salt Wash Member strata are known aquifers in other areas and probably have water yielding properties in Section 16. The Junction Creek and Entrada Sandstones are known aquifers and a well about three miles to the east in Section 13 probably taps one or both of these sandstones. Geologic units below the Entrada and above the Hermosa (Navajo, Wingate, Chinle) yield small quantities of water to wells in other areas, but nothing is known of their water yielding potential in the vicinity of the McElmo Dome Unit application", (Romero, 1985), [See appendix for depths to rock units penetrated by the Hovenweep wellbore).

The Leadville/Ouray formation water quality is quite variable. The TDS concentration ranges from 3,000 ppm to over 100,000 ppm depending where the water sample is taken. The lower TDS values are drawn from above the gas-water interface, where the water is in vapor form and drops out as condensation, with the change in pressure and temperature when the gas is produced. The higher TDS concentrations come from below the gas-water interface where "free water" is occurring. The injection zone, in the permit application, was shown to have a TDS concentration of 78,727 ppm. The injected fluid has a TDS concentration of 3,155 ppm, the porosity of the zone is 8% with a permeability of 2.7 millidarcy (md).

3. WELL CONSTRUCTION

The SWEPI Hovenweep (HWD-1) well was spudded November 21, 1978, and total depth (TD) was reached January 14, 1979. The well was initially drilled as a CO₂ exploratory well but the well was found to be uneconomical for CO₂ production. The well was originally perforated at 8289 - 8448 feet for testing productivity. On November 11, 1983, this interval was cement squeezed with 150 sacks of class H cement and production was abandoned. The well was re-perforated at 8580 -8534 feet on November 21, 1983 and stimulated with 125 bbls of 15% hydrochloric acid. An injectivity test and pressure test to 1000 psi were performed on December 2, 1983. It was completed for injection purposes on December 6, 1983.

ATTACHMENTS TO COMPLETION REPORT HWD-1

I. Geologic Information

- 1. Lithology and Stratigraphy
 - Description of Rock Units Penetrated A.

Name:

Dakota

Age:

Cretaceous

Depth:

Outcrops at surface

Thickness:

570'

Lithology:

Light gray sandstone and carbonaceous shale

Name:

Morrison Jurassic

Age: Depth:

570'

Thickness:

190'

Lithology:

Light Gray to pink sandstone and green or

red mudstone

Name:

Bluff

Age:

Jurassic

Depth:

760'

Thickness:

265;

Lithology:

Gray to buff sandstone

Name:

Summerville

Age:

Jurassic

Depth:

Thickness:

1025'

Lithology:

118' Red, sand mudstone, red sandstone and

minor chert

Name:

Entrada

Age:

Jurassic

Depth:

1143'

Thickness:

105'

Lithology:

Light buff, reddish brown or salmon

colored fine-grained sandstone

Name:

Carmel

Age:

Jurassic

Depth:

1248'

Thickness:

Lithology:

Reddish brown sandy siltstone and silty

sandstone

Name: Navajo Age: Jurassic

Depth: 1265' Thickness: 133'

Lithology: Buff to pale orange cross-bedded standstone

Name: Kayenta
Age: Triassic
Depth: 1398'
Thickness: 42'

Lithology: Reddish sandstone and interbeds of red or

green mudstone

Name: Wingate
Age: Triassic
Depth: 1440'
Thickness: 510'

Lithology: Pale orange and brown, fine-grained sandstone

Name: Chinle
Age: Triassic
Depth: 1950'
Thickness: 706'

Lithology: Reddish brown siltstone and sandstone,

reddish brown to variegated bentonitic mudstone, and minor gray conglomeratic

sandstone

Name: Moenkopi Age: Triassic Depth: 2656' Thickness: 49'

Lithology: Reddish brown laminated siltstone, sandy

siltstone and minor very fine-grained

sandstone

Name: Cutler Age: Permian Depth: 2705' Thickness: 1777'

Lithology: Interbedded red, orange, and purple

sandstones, siltstones and shales

Name: Honaker Trail Age: Pennsylvanian

Depth: 4482' Thickness: 1028'

Lithology: Interbedded gray sandstones, limestones

and shales with scattered amounts of gray

and orange chert.

Name:

Paradox

Age:

Pennsylvanian.

Depth:

5510' 2460'

Thickness: Lithology:

Interbedded salt, anhydrite, dark colored

dolomites and black shale

Name:

Pinkerton Trail Pennsylvanian

Age: Depth:

79701

Thickness:

146'

Lithology:

Light gray limestones and gray to graygreen shales, siltstones and sandstones

Name:

Molas

Age:

Pennsylvanian

Depth:

8116' 114'

Thickness: Lithology:

Interbedded red siltstones, sandstones,

light colored limestones and varicolored

shales

Name:

Leadville

Age:

Mississippian

Depth:

82301

Thickness:

284

Lithology:

Limestone, often oolitic and fossiliferous

changing to dolomite in the lower half

Name:

Ouray

Age:

Devonian

Depth:

85141

Thickness:

Lithology:

Limestone and dolomite with occasional

streaks of gray-green waxy shale

B. Description of Injection Unit

Name: Leadville Depth: 8230'

Thickness: 284'

Fluid Pressure: 2500 psi

Age: Mississippian

Porosity: (8%

2.7 md Permeability:

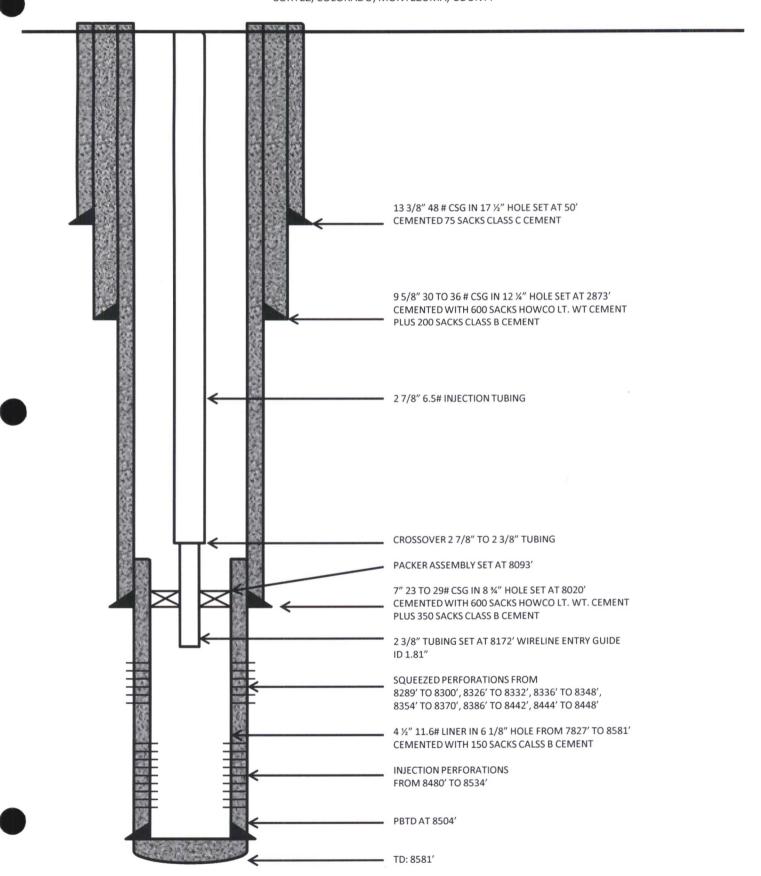
Bottom hole temperature: 180°F Lithology: Limestone and Dolomite

Bottom hole pressure: 3220 psi @ 3 BPM

Fracture pressure: 4330 psi

FIGURE 2-1

KINDER MORGAN CO2 COMPANY, LP HOVENWEEP HWD-1 EPA PERMIT NO. COS12ME-0042 CORTEZ, COLORADO, MONTEZUMA, COUNTY



A DIVISION OF ENSECO INCORPORATED

September 10, 1986

A. L. Pittman Shell Western E & P, Inc. P.O. Box 748 Lewis, CO 81327

Dear Larry:

Enclosed are the results for the analysis of the six water samples received August 6 and 7, 1986. Attached is Appendix A detailing the results for the compatibility study. Also enclosed is a list of standard methodologies and nominal detection limits. No satisfactory method is available for the analysis of glycol. TOX values may be suspect in some of the samples due to the high concentrations of inorganic chloride.

Please do not hesitate to call if you have any questions.

Sincerely,

Brian J. Rahn Project Coordinator

Inorganic Chemistry

BJR/JLP/bj Enclosures

RMAL #61862

Approved by

Jerry L. Parr

Technical Director

A DIVISION OF **ENSECO** INCORPORATED

September 11, 1986

A. L. Pittman Shell Western E & P, Inc. P.O. Box 748 Lewis, CO 81327

Dear Larry:

Enclosed is a revised report with the correct units for the Radiochemistry analysis. I apologize for any inconvenience this may have caused.

Please do not hesitate to call if you have any questions.

Brian J. Rahn

Project Coordinator Inorganic Chemistry

BJR/bi Enclosures

RMAL #61862

SAMPLE DESCRIPTION INFORMATION

for

Shell Western E & P, Inc.

RMA Sample No.	Sample Description	Sample Type	Date Sampled	Date Received
61862-01 61862-02	Hovenweep Disposal Water YA Produced Water	Water Water	08/04/86 08/04/86	08/06/86 08/06/86
61862-03	Yellow Jacket Facility HA Produced Water Hovenweep Facility	Water	08/04/86	08/06/86
61862-04 61862-05 61862-06	Moqui Produced Water Moqui Disposal Water Yellow Jacket Disposal Water	Water Water Water	08/05/86 08/05/86 08/06/86	08/06/86 08/06/86 08/07/86

September 11, 1986

ANALYTICAL RESULTS

for

Shell Western E & P, Inc.

 2.1 ± 1.3

 0.0 ± 5

 7.6 ± 5.8

 15 ± 8

RADIOCHEMISTRY		H - DBP		J. 710J.	
Parameter	Units	61862-01	6.1862-02	61862-03	61862-04
Radium 226 Radium 228 Gross alpha Gross beta	pCi/L pCi/L pCi/L pCi/L	13 ± 3 0.0 ± 4 23 ± 11 30 ± 9	4.1 ± 1 0.0 ± 4 15 ± 4 0.0 ± 6.8	160 ± 10 4.7 ± 5 340 ± 150 530 ± 170	9.4 ± 2.6 0.0 ± 4 21 ± 14 65 ± 21
Parameter	<u>Units</u>	61862-05	61862-06		

 7.5 ± 2.3

 0.0 ± 5

 21 ± 19

 97 ± 24

Radium 226

Radium 228

Gross alpha

Gross beta

pCi/L

pCi/L

pCi/L

pCi/L

ANALYTICAL RESULTS

for

Shell Western E & P, Inc.

INORGANIC PARAMETERS

Parameter	Units	618	62-01	618	62-02	618	62-03	618	62-04
Specific Conductance at 25°C Total Dissolved Solids Chloride Nitrate + Nitrite as N Sulfate Total Organic Carbon Total Organic Halogen Oil and Grease Fecal Coliform Total Coliform Hexavalent Chromium Specific Gravity at 80°F Compatibility*	units umhos/cm mg/L mg/L mg/L mg/L ugCl/L mg/L col./100 mL col./100 mL units	5.75 7160 4690 2130 ND 142 240 30 18 ND ND	(0.01) (1) (10) (3) (0.1) (5) (0.1) (5) (1) (1) (1) (0.01) (0.0001)	3.99 180 180 36 ND ND 12 5 ND ND 0.01 0.9968	(0.01) (1) (10) (3) (0.1) (5) (0.1) (5) (1) (1) (1) (0.01) (0.0001)	6.49 22800 13200 8440 0.2 836 15 180 ND ND ND 0.07 1.0107	(0.01) (1) (10) (3) (0.1) (5) (0.1) (5) (0.5) (1) (1) (0.001) (0.0001)	5.82 3680 2440 820 ND 146 49 7 2 ND ND ND 0.02 0.9977	(0.01) (1) (10) (3) (0.1) (5) (0.1) (5) (1) (1) (1) (0.01) (0.0001)

Parameter	Units	618	862-05	618	362-06
Specific Conductance at 25°C Total Dissolved Solids Chloride Nitrate + Nitrite as N Sulfate Total Organic Carbon Total Organic Halogen Oil and Grease Fecal Coliform Total Coliform Hexavalent Chromium Specific Gravity at 80°F Compatibility*	units umhos/cm mg/L mg/L mg/L mg/L ugCl-/L mg/L col./100 mL col./100 mL units	5.86 3570 2420 907 ND 125 51 460 47 ND ND ND ND	(0.01) (1) (10) (3) (0.1) (5) (0.1) (5) (1) (1) (1) (0.01) (0.0001)	5.00 1190 1000 281 ND 12 110 78 90 ND ND ND 0.04 1.0001	(0.01) (1) (10) (3) (0.1) (5) (0.1) (5) (0.5) (1) (1) (0.001)

^{*}See Appendix A

ND = Not detected. Detection limits in parentheses.

ANALYTICAL RESULTS

for

Shell Western E & P, Inc.

TRACE METALS

Chromium

Ma esium Potsium

Iron

Sodium

Strontium

Parameter	<u>Units</u>	613	862-01	618	862-02	618	862-03	618	362-04
Bamm Carium Chromium Iron Magnesium Potassium Sodium Strontium	mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L	0.020 89 0.58 10 28 21 148 0.14	(0.005) (0.1) (0.005) (0.05) (0.1) (0.3) (0.5) (0.005)	0.55 14 0.29 4.2 5.5 ND 3.6 0.035	(0.005) (0.1) (0.005) (0.05) (0.1) (0.3) (0.5) (0.005)	0.085 657 0.10 3.2 68 591 3980 4.2	(0.025) (0.5) (0.025) (0.25) (0.5) (1.5) (2.5) (0.025)	0.021 178 0.30 6.6 44 80 58 0.45	(0.005) (0.1) (0.005) (0.05) (0.1) (0.3) (0.5) (0.005)
Parameter	<u>Units</u>	618	62-05	618	62-06				
Barium Calcium Chromium	mg/L mg/L	0.038 296	(0.005) (0.1)	0.020 91	(0.005) (0.1)				

0.15

0.076

3.9

37

12

12

(0.005)

(0.05)

(0.1)

(0.3)

(0.5)

(0.005)

ND = Not detected. Detection limits in parentheses.

mg/L

mg/L

mg/L

mg/L

mg/L

mg/L

1.1

0.59

14

106

187

67

(0.005)

(0.05)

(0.1)

(0.3)

(0.5)

(0.005)

APPENDIX A

A DIVISION OF INCORPORATED

September 5, 1986

DH009

Jerry Parr Marketing Manager Rocky Mountain Analytical Lab 4955 Yarrow Street Arvada, Colorado 80002

Dear Jerry:

The enclosed report contains laboratory data concerning the disposal and production water samples from Shell Western E&P Inc. Compatibility and viscosity tests show no significant precipitate formation or viscosity increase during the test period. If you have any questions, please call

David W. Havis

Staff Research Scientist

DH/dp

Enclosure

cc: Maurice Jones

^{• 205} Alewife Brook Parkway, Cambridge, Massachusetts 02138 (617) 661-3111 Telex 650-256-7697 (MCI)

 ¹²³ Grove Avenue. Suite 118, Cedarhurst, New York 11516 (516) 295-1162

[•] do Bectech Trading Co., Ltd., P.O. Box 101-41, Taipei, Taiwan (R.O.C.) Tel. 5013908

ENSECO, INC. Southern Region

Subject

Production and disposal water samples from Shell Western E&P Inc. in Cortez, Colorado.

11

Conclusion

No significant precipitate formation or viscosity increase was observed during the combination of production and disposal water samples. The 30 ... of 75°F.

Shell Western E&P Inc. Page 2

Sample Identification

1.	MOQUI Produced Water	#61862-04	
2.	MOQUI Disposal Water	#61862-05	
3.	Hovenweep Produced Water	#61862-03	,
4.	Hovenweep Disposal Water	#61862-01	
5.	Yellow Jacket Produced Water	#61862-02	
6.	Yellow Jacket Disposal Water	#61862-06	

Testing Procedure

- 1. Each sample acclimated to a room temperature of 75°F.
- Before the combination of production and disposal water sample pairs, precipitate formation and individual water temperatures were recorded.
- 3. After mixing, sample monitoring of precipitate formation, viscosity change, and temperature fluctuations occurred at 15 minute intervals.

..:

-:/-

Test Data

Table 1: Shell Western E&P Inc. Production and Disposal Water Samples

				Jan	h 162
Sample Identification	Initial Sample Volume mls.	Initial Sample Temperature °F.	Precipitate Formation	Turbidity	
MOQUI Produced Water 61862-04	150	75	None		Color
MOQUI Disposal Water 61862-05	150		Slight,	None	None
Hovenweep Produced Water 61862-03		75	Yellow-brown Slight,	Slight	Yellow
Hovenweep Disposal Water	150	75	Yellow-brown	None	Yellow
61862-01 Yellow Jacket Produced Water	150	75	Slight, White	Slight	White
61862-02 Yellow Jacket	150	75	None	None	None
Disposal Water 61862-06	150	75	Slight, White	Slight	White

Table 2: Combination of Production and Disposal Sample Pairs

				, 4113	
Sample Identification	Final Sample Volume mls.	Final Sample Temperature °F.	Precipitate Formation	-	
MOOLIT			rormation	Turbidity	Color
MOQUI	300	75	None	Slight	Yellow
Hovenweep .	300	75	M	3	ICITON
		75	None	None	Clear
Yellow Jacket	300	75	None	Slight	
				Siight	White

Shell Western E&P Inc. Page 4

Test data shows no apparent compatibility or viscosity problems with the three pairs of production and disposal waters.

DWH/dp

111

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Discussion

Inorganic Analytical Methodology

Parameter	Units	Nominal Detection Limit ^a	Methodology R	eference	Preservation Bottle No.	Maximum HoldingTime
MAJOR IONS						
Sodium	mg/l	0.5	ICP Emission Spectroscopy	3	4	6 months
Potassium	mg/l	0.3	ICP Emission Spectroscopy	3	4	6 months
Calcium	mg/l	0.1	ICP Emission Spectroscopy	3	4	6 months
Magnesium	mg/l	0.1	ICP Emission Spectroscopy	3	4	6 months
Chloride	mg/l	3	Manual Titrimetric, Hg (NO ₃) ₂ Automated Colorimetric	•	3 1	28 days
Flore: Je	/1	0 1	Ferricyanide	1-325.2	1	28 days
Fluoride	mg/l	0.1	Electrode	1-340.2/2-4138		28 days
Sulfate	mg/I	5	Manual Turbidimetric	1-375.4/2-4260		28 days
Total Alkalinity on CaCO			Automated Colorimetric MTB	1-375.2	1	28 days
Total Alkalinity as CaCO ₃ at pH 4.5 Carbonate Alkalinity as CaCO ₃	mg/l	5	Titrimetric	1-310.1/2-403	1	14 days
at pH 8.3 Bicarbonate Alkalinity as CaCo	mg/l	5	Titrimetric	1-310.1/2-403	1	14 days
at pH 4.5	³ mg/l	5	Titrimetric	1-310.1/2-403	1	14 days
Hydroxide Alkalinity as CaCO		5	Calculation	2-403	_	-
Nitrate+Nitrite as N	mg/l	0.1	Manual Cd Reduction -			
	Ü	0.1	Colorimetric Automated Cd Reduction -	1-353.3/2-4180	2	28 days
			Colorimetric	1-353.2	2	28 days
Total Cations	meq/I	0.1	Calculation	2-104C	-	_
Total Anions	meq/I	0.1	Calculation	2-104C	-	_
Difference	%	0.1	Calculation	2-104C	-	-
RADIOCHEMISTRY						
Gross Alpha	pCi/I	0.1	Proportional Counter	2-703	5	6 months
Gross Beta	pCi/I	0.1	Proportional Counter	2-703	5	6 months
Radium 226	pCi/I	0.1	Separation - Counter	2-705	5	6 months
Radium 228	pCi/I	0.1	Separation - Counter	2-707	5	6 months
Uranium	mg/l	0.005	Fluorimetric	4-D2907-75	5	6 months
	9			~ - / - / /		- 1110111111

Inorganic Analytical Methodology (Continued)

Parameter	Units	Nominal Detection Limit ^a	Methodology R	Reference	Preservation Bottle No.	Maximum Holding Time
TRACE METALSC						
Aluminum	mg/l	0.05	ICP Emission Spectroscopy	3	4 .	6 months
Antimony	mg/l	0.002	Furnace Atomic Absorption	1-204.2	4	6 months
Arsenic	mg/I	0.002	Furnace Atomic Absorption	1-206.2	4	6 months
Barium	mg/l	0.005	ICP Emission Spectroscopy	3	4	6 months
Beryllium	mg/l	0.001	ICP Emission Spectroscopy	3	4	6 month
Boron	mg/l	0.004	ICP Emission Spectroscopy	3	4	6 moles
Cadmium	mg/l	0.002	ICP Emission Spectroscopy	3	4	6 months
Chromium	mg/l	0.005	ICP Emission Spectroscopy	3	4	6 months
Cobalt	mg/l	0.003	ICP Emission Spectroscopy	3	4	6 months
Copper	mg/l	0.002	ICP Emission Spectroscopy	3	4	6 months
Iron	mg/I	0.05	ICP Emission Spectroscopy	3	4	6 months
Lead	mg/l	0.025	ICP Emission Spectroscopy	3	4	6 months
	0.	0.001	Furnace Atomic Absorption	1-239.2	4	6 months
Manganese	mg/l	0.005	ICP Emission Spectroscopy	3	4	6 months
Mercury	mg/l	0.0002	Cold Vapor Atomic Absorption		4	6 months
Molybdenum	rng/l	0.005	ICP Emission Spectroscopy	3	4	6 months
Nickel	mg/l	0.01	ICP Emission Spectroscopy	3	4	6 months
Selenium	mg/1	0.002	Furnace Atomic Absorption	1-270.2	4	6 months
Silver	mg/I	0.003	ICP Emission Spectroscopy	3	4	6 months
Strontium	mg/I	0.005	ICP Emission Spectroscopy	3	4	6 months
Thallium	mg/I	0.002	Furnace Atomic Absorption	1-279.2	4	6 month
Tin	mg/l	0.03	ICP Emission Spectroscopy	3	4	6 moles
Titanium	mg/l	0.002	ICP Emission Spectroscopy	3	4	6 months
Vanadium	mg/l	0.002	ICP Emission Spectroscopy	3	4	6 months
Zinc	mg/l	0.004	ICP Emission Spectroscopy	3	4	6 months
INORGANIC PARAMETERS						
рН	units	0.01	Meter	1-150.1; 2-42	3 1	ASAP
Specific Conductance at 25°C	umhos	/cm 1	Bridge	1-120.1; 2-205		28 days
Total Dissolved Solids	mg/I	10	Gravimetric, 180°C	1-160.1; 2-209	9B 1	7 days
Total Suspended Solids	mg/I	2	Gravimetric, 105°C	1-160.2	1	7 days
Total Solids	mg/l	10	Gravimetric, 105°C	1-160.3	1	7 days
Total Volatile Solids	mg/l	10	Gravimetric, 550°C	1-160.4	1	7 days
Ortho-Phosphate as P	mg/I	0.01	Single Reagent Colorimetric	1-365.2; 2-424	F I	48 hours

Inorganic Analytical Methodology (Continued)

Parameter	<u>Units</u> <u>De</u>	Nominal etection Limit a	Methodology R	eference	Preserva Bottle	Maximum HoldingTime
INORGANIC PARAMETERS (Continued)						
Total Phosphorus as P	mg/l	0.06	Digestion; ICP Emission Spectroscopy	1-4.1.4;	3 4	28 days
Silica as SiO ₂	mg/l mg/l	0.01 0.1 1	Digestion - Colorimetric ICP Emission Spectroscopy Colorimetric	1-365.2; 3 1-370.1;	1-424C,F 2 4 2-425C 1	28 days 28 days 28 day
Biological Oxygen Demand Chemical Oxygen Demand	mg/l mg/l	2 5	Dilution Bottle - D.O. Probe Micro Colorimetric	1-405.1;	2-507 1 2-508A 2	48 hours 28 days
Total Organic Carbon Ammonia as N	mg/l mg/l	0.1 0.1 0.1	Oxidation-Infrared Absorption Electrode Automated Colorimetric	1-415.1; 1-350.3; 1-350.1		28 days 28 days 28 days
Total Kjeldahl Nitrogen as		0.1	Digestion - Electrode Digestion - Colorimetric	1-351.4; 1-351.2	2-420B 2 2	28 days 28 days
Total Organic Nitrogen as N Oil and Grease Free Cyanide	N mg/l mg/l mg/l	0.1 1 0.01	Calculation (TKN - NH ₃) Freon Extraction-Gravimetric Chlorination-Distillation-	1-413.1;	2-503A 3	28 days
Total Cyanide	mg/l	0.01	Colorimetric Distillation - Colorimetric Distillation - Colorimetric	1-335.2;	2-412F,D 6 2-412B,D 6 2-510A,B 2	14 days 14 days 28 days
	mg/l onies/100 ml onies/100 ml	1	Membrane Filter Membrane Filter	2-909C 2-909A	8 8	ASAP ASAP
Bromide Residual Chlorine	mg/l mg/l	0.1	Colorimetric Amperometric	2-405 1-330.2;		28 da ASAP 24 hours
Hexavalent Chromium Color Hardness as CaCO ₃	mg/l units mg/l	0.01 5 5	Colorimetric Pt-Co Colorimetric Calculation	1-218.4; 1-110.2; 2-314A		48 hours 6 months
Nitrite as N Sulfide Sulfite	mg/l mg/l mg/l	0.01 0.05 2	Colorimetric Titrimetric - Electrode Titrimetric	1-354.1; 1-376.1; 1-377.1;	2-427B,D 7	48 hours 7 days ASAP
MBAS (Surfactants) Turbidity	mg/I NTU	0.1 0.1	Colorimetric Turbidimeter	1-425.1;	2-512A 1	48 hours 48 hours

Inorganic Analytical Methodology (Continued)

References

- (1) "Methods for Chemical Analysis of Water and Wastes", EPA-600/4-79-020, EMSL, Cincinnati, 1979.
- (2) "Standard Methods for the Examination of Water and Wastewater", 15th Edition, APHA, 1980.
- (3) 40 CFR 136.3, Table II.
- (4) "Annual Book of ASTM Standards", Part 31, Water, 1980.

Notes

^aNominal values are the best achievable with the listed analytical method. Interferences in specific samples may result in a higher detection limit.

^bApplicable to NPDES wastes as updated by Robert C. Booth, Director, EMSL-Cincinnati, September 22, 1981.

^cDigestion procedure 1-4.1.4 used for elements determined by ICP Emission Spectroscopy when determining total metals. Digestion procedures for graphite furnace elements included with reference listed.

6/5/86

GUIDELINES FOR SAMPLE BOTTLES AND PRESERVATIVESA, b

Bottle No.	$\frac{\text{Parameters}}{\text{C1}^{-},\text{F}^{-},\text{SO}_{4}^{-},\text{Tot. Alk., CO}_{3}^{-}\text{Alk., HCO}_{3}^{-}\text{Alk.,}}$	<u>Container</u> 500 mL poly	Preservative 4°C	Notes Provide unfiltered
	SiO_2 , BOD, Br ⁻ , res. $C1_2$, Cr^{+6} , color, NO_2^- , $SO_3^=$,			sample for solids and turbidity.
	OH Alk., pH, spec. cond., TDS, TSS, TS, TVS, g-PC	04		
	MBAS, Turbidity.			
2	Tot.P, COD, TOC, NH ₃ , TKN, TON, Phenolics $NO_3 + NO_2$.	1 liter glass	2 ml 50% H ₂ SO ₄ , 4°C	
3	O & G	1 liter glass	4 ml 50% H ₂ SO ₄ , 4°C	Do not filter, collect directly in bottle.
4	Na, K, Ca, Mg, A Sb, As, Ba, Be, B, Cd, Cr, Co, Cu, Fe, Pb, Mn, Hg, Mo, Ni, Se, Ag, Sr, Tl, Sn, Ti, V, Zn, ICP, Hardness	500 ml poly	5 ml 50% HNO ₃	Provide separate samples for total and dissolved sample (filter before adding to bottle).
5	Alpha, Beta, Ra ²²⁶ , Ra ²²⁸ , U	1 liter poly (no Ra ²²⁸) ½ gallon poly (with Ra ²²⁸)	10 ml 50% $\mathrm{HNO_3}$ 20 ml 50% $\mathrm{HNO_3}$	
6	Free CN, Tot. CN	500 ml poly	2 ml 50% NaOH, 4°C	
7	Sulfide	250 ml poly	1 ml 1 N Zn acetate, 1 ml 50% NaOH, 4 ^o C	
8	Fecal coli., total coli.	8 oz. sterile	4°C	Collect directly in sterile bottle.
11	VOA, purgeable organics, THM	3-40 ml glass vial	4°C	Completely fill bottle, leave no air bubbles.
12	B/NA	1 liter glass	4°C	
13	Pest./PCB	1 liter glass	4°C	
14	Herbicides	1 liter glass	4°C	
15	TOX	1 liter glass	4°C	

⁸40 CFR 136.3 Table II.

b Note: Certain other non-routine samples may require different perservation to remove interferences as specified in 40 CFR 136.3 Table II.